Graphene production at industrial scale



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Abstract

Graphene is a two-dimensional (2D) allotrope of carbon; effectively a planar and atomically thin sheet of graphite. In this 2D form, graphene exhibits remarkable thermal and electrical properties with an unprecedented surface-to-volume ratio enhancing sensitivity. As expected, the industrial production, transfer and handling of 2D graphene is challenging. Here, challenges unique to the industrial scale production of graphene are presented: the scientific/technical solutions were guided by maximizing graphene material quality under the constraint of minimized cost.

Introduction & Method

Pristine, atomically thin graphene is produced using *atmospheric pressure* Chemical Vapor Deposition (CVD) on copper (Cu) foil substrates. Industrial, cost-effective throughput requires a continuous growth process. Fortunately, the roll-to-roll continuous method of production is compatible with Cu foil substrate required for monolayer graphene growth. At General Graphene, we have adapted CVD to the roll-to-roll method. In doing so, the advantage of working with well defined gas concentrations, in a fixed volume, is lost, which is a key strength afforded by CVD in quartz tubes (see below). Furnace

Time-controlled method: Quartz tube (*static substrate, low throughput*) Graphene growth occurs in two basic stages; (1) Cu foil annealing then (2) graphene growth. Fixed volumes of precursor gases, e.g., Ar, H_2 and CH_4 , flow at constant concentration for a fixed time, then exchanged.







t_o: Cu foil annealing



t_{1b}: graphene growth (final)





Position-controlled method: Furnace duct (*moving substrate*, *high throughput*) Graphene growth occurs using the same annealing + growth process but the substrate moves through a furnace with variable gas concentration in the rolling direction (x)



Results

Single layer graphene is produced by CVD using following parameter ranges Recipe

- Cu foil substrate $(20 70 \mu m)$
- 2%H₂/Ar @ 10¹–10² [L/min]
- 1000 ppm CH₄/Ar @ 10¹ 10² [L/min]
- $T \ge 10^3 [C]$
- t = 600 [cm] / 20 [cm/min] = 30 [min] ...where 720 [cm] is the furnace CVD length

Electron image

A single graphene grain viewed looking down on Cu foil (dark hexagon). The lower contrast (greyish) background is the texture of the underlying Cu surface. The symmetry of the grain edge indicates a single grain of commonly oriented carbon atoms

Cu ...near melting temperature

Additional Growth Recipe Features

- Methane (CH_4) concentration must increase, or must be sufficient, to overcome the reduction in exposed catalytic Cu surface area as the graphene layer approaches full coverage
- Hydrogen (H₂) serves a dual role of promoting CH_4 dehydrogenation while also chemically reducing any copper oxides at the surface, i.e., Cu₂O and/or CuO

Results ... before CVD

As received Cu foil substrate must be electropolished to reduce the surface roughness as excessive roughness degrades graphene quality leading to a defective monolayer. Scanning laser profilometry is used to assess the Cu foil roughness before and after electropolishing.



Laser scanning profilometry

A surface roughness range of 1 um was required to encapsulate all Cu foil surface roughness spanning an area of 280 µm x 210 µm while, after roll-to-roll electropolishing in 85 wt% H₃PO₄, the roughness range to encapsulate all surface roughness, over the same area, was reduced to 500 nm, a value acceptable for subsequent annealing

Final Graphene Transfer

The process flow as summarized above is amenable to multiple pathways of graphene transfer. Graphene "transfer" consists of;

Resultsafter CVD

02

-0-2

-04

Raman microscopy is an invaluable, nondestructive tool for quantification of graphene nanostructure providing information on the defect density as well as excessive CVD growth where multiple layers of graphene may deposit



Transfer (option 1) \rightarrow roll-to-roll: graphene on flexible polymer

- 1. Application of a polymeric support to the exposed surface of graphene
 - Poly (methyl methacrylate) (PMMA) [spin coating², spray coating^{1,2}]
 - Polyethylene terephthalate/ethylene vinyl acetate (PET/EVA) [film lamination^{1,2}]
- 2. Original substrate removal
 - Wet etching^{1,2}
 - Oxidative^{1,2}
 - Electrochemical^{1,2}
- Transfer to new substrate² 3.
 - Polymer dissolution/cleaning (1,2) red numbers refers to pathway options shown at the right